

# Waste Gasification by Thermal Plasma: A Review

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**Abstract** This paper proposes an overview of waste-to-energy conversion by gasification processes based on thermal plasma. In the first part, basic aspects of the gasification process have been discussed: chemical reaction in gasification, main reactor configuration, chemical conversion performances, tar content in syngas and performances in function of the design and the operation conditions (temperature, pressure, oxidizing agent...). In the second part of the paper are compared the performances, available in the scientific literature, of various waste gasification processes based on thermal plasma (DC or AC plasma torches) at lab scale versus typical performances of waste autothermal gasification: LHV of the syngas, cold gas efficiency and net electrical efficiency. In the last part, a review has been done on the various torch technologies used for waste gasification by plasma at industrial scale, the major companies on this market and the perspectives of the industrial development of the waste gasification by thermal plasma. The main conclusions are that plasma technology is considered as a highly attractive route for the processing of waste-to-energy and can be easily adapted to the treatment of various wastes (municipal solid wastes, heavy oil, used car tires, medical wastes ...). The high enthalpy, the residence time and high temperature in plasma can advantageously improve the conditions for gasification, which are inaccessible in other thermal processes and can allow reaching, due to low tar content in the syngas, better net electrical efficiency than autothermal processes.

**Keywords** *Allothermal process; Gasification; Syngas; Thermal plasma torch; Waste-to-Energy; tar content*

## Introduction

Since the end of World War II, all developed countries generate more and more domiciliary and industrial wastes per capita at a level that is becoming unmanageable, causing permanent damages to the environment. For example, in Japan, it is estimated that each inhabitant produces around 1.1 kg/day of Municipal Solid Waste (MSW) and the total quantity is about  $5.2 \times 10^7$  tons/yr. Moreover, total quantity of industrial waste is about  $4 \times 10^8$  tons/yr [1].

Public and political awareness to environmental issues have led to plan to implement strategies for waste management. In parallel, the energy consumption continuously grow in the world due to the increasing population, the industrial development and the consumerism which has become a life standard in industrial countries.

The sustainable strategy for the waste management is to improve waste treatment in the aim to reduce their landfill disposal and minimize the environmental impact. For few years, wastes became one of the renewable resources that could play a major role in renewable energy [2]. Various thermal processes, like combustion, pyrolysis or gasification have been developed for treating these wastes in the aim to recover energy from the organic fraction [3-22]. Various thermal processes, like combustion [3-6], pyrolysis [3-7] or gasification [3-22], have been developed for treating wastes in the aim to recover energy from the organic fraction. In these papers, which are mainly reviews on the thermo-chemical conversion of biomass/waste to energy, are discussed the performances of the existing thermal processes, and more particularly the gasification processes, in function of the technologies used. However, waste gasification by thermal plasma, which is a relatively new technology in the field of waste treatment by gasification, is often forgotten or neglected in the literature and only few recent studies have analyzed the plasma process as a solution for Waste-to-Energy recovery [3, 4, 6, 8-10]. However, in these papers, the plasma technologies are often partially studied, so the main purpose of this present paper is to focus on the thermal plasma technologies for the treatment of municipal and industrial wastes for energy recovery.

There are numerous wastes with an organic content which may be suitable for gasification or other thermochemical processes. MSW is a heterogeneous fuel containing a very wide variety of solid wastes. Due to the presence of some post-recycling materials, such as paper fiber and plastics, its heating value can be high and gasification proposes to take advantage of this. The chemical composition of MSW can be compared to any solid organic fuel like coal or biomass. According to [23], the element composition of MSW is in the range (Weight %): C – (17 – 30), H<sub>2</sub> – (1.5 – 3.4), O<sub>2</sub> – (8 – 23), H<sub>2</sub>O – (24 – 34), ashes – (18 – 43) and the average

specific combustion heat of MSW is in the range from 5 to 10 MJ.kg<sup>-1</sup>. In [24, 25], thermal plasma pyrolysis of old tires has been tested and the combustion heat value of the produced gas was in the range 4 to 7 MJ.Nm<sup>-3</sup> in the first one and 5.3 to 7.9 MJ.Nm<sup>-3</sup> in the second one (up to 9 MJ.Nm<sup>-3</sup> with water gas shift reaction).

This paper reviews the current status of thermal plasma technologies for the treatment of domiciliary and industrial wastes for energy recovery. The inorganic waste plasma treatments by melting and vitrification are not discussed here and are not addressed in this review [26-31].

In the current context of conventional fossil resource depletion, global warming and rising waste, gasification of wastes appears as an interesting alternative compared to combustion processes. Indeed, the usual methods based on the incineration of wastes are low energy balances for electricity production. Net electrical efficiencies from 18 % to 22 % can theoretically be achieved at an industrial scale, resulting from the use of a boiler associated with a steam turbine [16]. In opposition, gasification by thermochemical decomposition of organic material allows the production of synthesis gas, i.e. syngas, in which one can recover up to 80 % of the chemical energy contained in the organic matter initially treated. Based on these performances, a plasma gasifier associated with a gas turbine combined cycle power plant can target up to 46.2 % efficiency [32]. Moreover, this synthesis gas produced by gasification, mainly composed of CO and H<sub>2</sub> can also be used as feedstock for the production of synthetic liquid fuels in processes such as Fischer-Tropsch process.

However, conventional methods based on autothermal gasification present some limitations that might be overcome through plasma [33], particularly in terms of: material yield, syngas purity, energy efficiency, dynamic response, compactness and flexibility. Injected plasma power can be adjusted independently of the heating value of the treated material.

On the chemical aspect, the thermal plasma can advantageously contribute to the gasification by accelerating the kinetics and improving high temperature cracking of impurities in the syngas produced. On the thermal aspect, enthalpy provided by the plasma can easily be adjusted by the tuning of the electrical power supplied to the system, making the process independent of the ratio O/C and the nature of the plasma medium (neutral, oxidizing or reducing atmosphere), contrary to the autothermal gasification processes.

## **Gasification process**

Gasification process was discovered in 1699 by Dean Clayton [34]. It was implemented during the nineteenth century in factories for producing town gas. The first gas plant was established in 1812 in London. With the discovery of the Fischer Tropsch Process in 1923 by Franz Fischer and Hans Tropsch, it became possible to convert coal to liquid fuel. During World War II, German army needed to improve the use of the gasification process for fuel and chemical production. The end of the war and the availability of cheap fossil fuel reduced the usefulness of this process but with the current context of conventional fossil resource depletion and the increasing of fuel prices, gasification of wastes appears as an interesting alternative for energy.

Gasification is an incomplete oxidation of organic compounds after a pyrolysis decomposition step. The oxygen contained in the oxidizing agent used for the gasification (Air, oxygen, CO<sub>2</sub> or steam water) reacts with carbon to achieve a combustible gas, called “syngas”. This syngas is mainly composed of carbon monoxide (CO) and hydrogen (H<sub>2</sub>) with low quantities of carbon dioxide (CO<sub>2</sub>), water (H<sub>2</sub>O), methane (CH<sub>4</sub>), hydrogen sulfide (H<sub>2</sub>S), ammonia (NH<sub>3</sub>), and under certain conditions, solid carbon (C), nitrogen (N<sub>2</sub>), argon (Ar) and some tar traces. Nitrogen and argon coming from the use of air as the reactant or are due to their use as plasma gas.

This synthesis gas produced can be used as feedstock in: (i) Fischer-Tropsch process for liquid fuel production [35], (ii) gas turbine or fuel cell for electricity production [36-38] or (iii) chemicals products as ammonia, methanol and hydrogen [39].

### **Chemical reaction in gasification**

The waste conversion into syngas involves complex chemical reactions. Heterogeneous reactions take place in gas-solid phase while the homogeneous reactions occur in gas-gas phase. The main chemical reactions of gasification occurring after the pyrolysis of the wastes are given as followed in Table 1 [3, 9]:

The homogeneous reactions (reactions 9-12) are almost instantaneous in high temperature conditions in contrast to heterogeneous reactions (reactions 1-8).

A very large number of gasification reactions take place in the reactor but we can differentiate three of them which are independent gasification reactions: Water-gas reaction (6), Boudouard reaction (7) and hydrogasification (8). In the gas phase, these reactions can be reduced to only two: Water-gas shift reaction (11) which is the combination of the reactions (6) and (7) and methanation (12) which is the combination of the reactions (6) and (8).

It is important to notice that all these gasification reactions, except the oxidation ones, are equilibrium reactions. The final composition of the crude syngas will be determined by reaction rates and also by the effect of catalysts which is important for tar decomposition in the reactor, rather than by the thermodynamic equilibrium [9].

### **Main reactor configuration**

There are presently several tens of different gasification processes which differ by the configuration of the reactors. These different configurations are fully described in numerous books and scientific papers [8, 36, 40, 41]. The main different reactor configurations are: Downdraft Fixed Bed, Updraft Fixed Bed, Bubbling Fluidized Bed, Circulating Fluidized Bed, Entrained Flow, Rotary Kiln, Moving Grate... These different configurations have been analyzed and commented in [9] and the main conclusions are as follows: It is commonly accepted that the three main reactor configurations are updraft, downdraft and fluidized bed.

Reed [42] gives the following description for each configuration:

In **updraft gasifiers**, the wastes are fed by the top of the reactor. The oxidizing agent, which may be air, oxygen, CO<sub>2</sub> or steam, is fed by the bottom of the reactor. The gasification reaction takes place in the bottom of the reactor between the downcoming material and the ascending gas. The reaction temperature is between 1 300 K and 1 700 K. The rise of the hot gas starts waste pyrolysis at lower temperatures and dries it. The tar levels in the crude gas with this reactor configuration are between 10 % and 20 %, which makes them difficult to clean for electricity applications.

In **downdraft gasifiers**, the wastes are fed by the bottom part of the reactor with the oxidizing agent, which may be air, oxygen, CO<sub>2</sub> or steam. The major part of the tars is burned for the pyrolysis of the wastes. This process is called "flaming pyrolysis". Thus, the tar levels in this reactor configuration are very low, around 0.1 %, as the major part of tars is burned to supply the energy for the pyrolysis / gasification reactions of the wastes. This reactor configuration is particularly suitable for the production of clean gas requiring low post-treatment for their use in electricity production with gas turbines. However, the operation generally requires a long residence time (1 h to 3 h) [41]. This configuration is considered most attractive to small units of 80 kWe-500 kWe and has the disadvantage to have low energy efficiency but with low tar concentrations [34].

In the case of **fluidized bed gasifiers**, the oxidizing gas allows the suspension of the treated waste. In this configuration, there is a mixture of the two phenomena identified previously in

the updraft and downdraft reactors. Thus, the tar rate is at an intermediate level between the updraft and downdraft reactors, between 1 % and 5 %. Gasification reactions are homogenized by suspended grounded wastes in the reactor. This method optimizes the temperature along the reactor and has a high reaction rate for short residence time (less than 30 min). The disadvantage of this configuration is the high proportion of particulates (tars) in the exhaust gas that requires high gas treatment and has low mass and energy yields [34].

### **Chemical conversion performances**

Different criteria are frequently quoted for gasification processes. In order to compare different processes, we define energy efficiency (also called cold gas efficiency),  $H_2$  rate and CO rate as follows (Equations 14 to 16). Cold gas efficiency is the energy produced by syngas combustion divided by the energy produced by direct combustion of product incremented by the added energy (electric or fuel) for allothermal processes. This efficiency does not take into account the steam consumption and electricity (related to pure oxygen production), or heat recovery by cooling synthesis gas (steam).

Fuel gas production is the flow of the gas mixture produced by gasification per kilogram of product treated in the reactor. When air is used as oxidant in the reactor, we can use, in this particular case, the formula [15]:

$$Fuel\ gas\ production\ (Nm^3.kg^{-1}) = \frac{air\ flow\ rate\ (Nm^3.s^{-1}) \times 0.79}{[1 - (CO + CO_2 + H_2 + CH_4 + C_2H_2)/100] \times feeding\ rate\ (kg.s^{-1})} \quad (13)$$

In (13), fuel gas production is function of the ratio of the nitrogen at the entrance of the process to the nitrogen in the mixture produced. In this particular case, which cannot be applied for all gasification situations, it is assumed that the conversion is total (no oxygen gas in the crude gas) and the only gases produced during gasification are CO, CO<sub>2</sub>, H<sub>2</sub>, CH<sub>4</sub> and C<sub>2</sub>H<sub>2</sub>. In this case, the waste used as feedstock is only composed of C and H (no chemical species like S, Cl ...).

### **Energy efficiency**

Energy efficiency of the process (or cold gas efficiency) is defined by the ratio of the Lower Heating Value (LHV) of cold gas to the LHV of the waste treated, incremented by the added

energy (electric or fuel) for allothermal processes per kg of waste. It is defined by the following expression:

$$\eta = \frac{LHV \text{ of cold gas } (kJ.Nm^{-3}) \times \text{fuel gas production } (Nm^3.kg^{-1})}{LHV \text{ of waste treated } (kJ.kg^{-1}) + \text{allothermal Power } (kW) / \text{waste flow rate } (kg.s^{-1})} \quad (14)$$

In the particular case of waste gasification by thermal plasma, we have to take into account the origin of the electric energy consumed to create the plasma. If the electric energy comes from the electric energy generated by the process, the allothermal power is equal to the electric energy consumed to create the plasma. If the electric energy comes from a primary thermal power plant, the Allothermal Power =  $P_{\text{Plasma(electrical)}} / (\text{Conversion efficiency of the thermodynamic cycle -Carnot-})$ . Generally, the conversion efficiency of thermal power plant is between 30 % and 40 % for a single cycle steam power plant and can be up to 60 % for a Combined Cycle Gas Turbine (CCGT) power plant.

### ***H<sub>2</sub> and CO yields***

H<sub>2</sub> yield is defined by the ratio of the mass of hydrogen in the syngas produced per the mass of hydrogen introduced. For the CO yield, it is the ratio of the mass of carbon atoms in the CO produced per the mass of the carbon atoms injected. These ratios are given by the formulas [43]:

$$H_2 \text{ yield} = \frac{H \text{ atoms in the syngas}}{H \text{ atoms injected}} \quad (15)$$

$$CO \text{ yield} = \frac{C \text{ atoms in the formed CO}}{C \text{ atoms injected}} \quad (16)$$

It is important to keep in mind that the H<sub>2</sub> rate in the crude syngas is strongly linked to the oxidizing agent injected and/or the moisture content in the waste treated. As H<sub>2</sub> yield is not representative of the conversion rate of the processes, only the CO yield can be used to provide good information on the mass balance and on the performances of the gasification processes.

### **Tar content in syngas**

This new interest for gasification was accompanied by an expansion of devices based on the syngas exploitation associated with waste gasification, like internal combustion engines (ICEs), gas turbines or fuel cells.

Tar presence in syngas is the most problematic parameter in any industrial gasification processes. This could have important implications in the design and the operation of gasifiers to ensure adequate control of reaction conditions. These tar constituents can be used as indicators of overall reactor performance and design [44].

Tars are complex mixture of condensable hydrocarbons or organic compounds having a molecular weight higher than benzene  $C_6H_6$ . This definition was introduced by the tar protocol measurement at the IEA Gasification Task meeting at Brussels in 1998 [45]. The tar rate is representative of the quantity of tars mixed with the syngas after gasification of the organic material ( $g.m^{-3}$ ).

The differences in tar nature and quantities are mainly function of the processing conditions, the applied technology and the nature of the wastes to be treated.

In his survey of biomass gasification, Reed [42] already concluded in 2001 that the Achilles heel of biomass gasification is the amount of tars contained in the syngas produced (0.1 – 10 %) and the technical feasibility and economical viability of biomass gasifier at an industrial scale will be strongly linked to the performances of the cleaning processes.

Depending on applications focused, tar concentrations in the syngas have to be mastered or cleaned. The scientific literature contains many data on the tar reduction, conversion and/or destruction in waste gasification processes. More than 400 papers have been referenced by Milne [46].

They focused on tar removal through physical processes and “tar” conversion through thermochemical and catalytic processes (Thermal, steam, partially oxidative, catalytic and/or plasma processes). The choice of the cleaning process depends specifically on the applications referred.

### ***Tar levels from gasifiers***

The results reported from the literature for tar rates, from the three main categories of gasifiers, are summarized in Table 2, showing a wide range of values, within each case (updraft, downdraft and fluidized bed) some ranges spanning from one to two orders of magnitude [46].



There is a general agreement on the relative order of magnitude of tar rates in function of the gasification process. They can be classified as follows: updraft gasifiers being the “dirtiest”, downdraft the “cleanest” and fluidized beds intermediate with an average value for updraft reactor at  $100 \text{ g.Nm}^{-3}$ , fluidized beds at  $10 \text{ g.Nm}^{-3}$  and downdraft reactor at  $1 \text{ g.Nm}^{-3}$  [46].

Any kind of material can be used as feedstock if it contains a certain amount of organic material inside (MSW, used tires, paper mill waste, plastic waste, liquid and solid hazardous waste, Refuse Derived Fuel (RDF, i.e. mixture of plastics, paper, wood and dried organic material), medical waste, biomass wastes ...).

The choice of the gasifier technology will depend of several fuel requirements like the particle size, the morphology, the moisture content, the ash content, the ash melting point, the bulk density, the temperature profile in the gasifier, the heat exchange, the residence time, the conversion efficiency, the process flexibility... The limitations and the categories of the materials used as feedstock in gasifier have been already studied and discussed in several papers, more particularly in an excellent review by Arena [9] in which are well summarized all the fuel requirements in function of the gasifier technologies.

### ***Tar tolerance of End-Use Devices***

A very important topic is the tar tolerance of the end-use devices. Many data are available from R&D activities and from field experience, mainly coming from manufacturers.

Depending on the applications referred for energy and chemicals production, the tar tolerance thresholds estimated are listed below [46]:

- Stirling Engines and turbines can work with raw gas (External combustion). No cleaning of the syngas is necessary for these applications but the energetic yields of these devices are low (around 20 %).
- Compressors accept a tar limit between  $100 \text{ mg.Nm}^{-3}$  and  $500 \text{ mg.Nm}^{-3}$ . This option is interesting for the syngas storage but depending on the end-use of this gas, a subsequent processing of the syngas for cleaning will be mandatory.
- The internal combustion engines tolerate a maximum tar concentration of  $50 \text{ mg.Nm}^{-3}$  for the lightest compounds and  $5 \text{ mg.Nm}^{-3}$  for the heaviest, and a concentration in solid particles lower than  $30 \text{ mg.Nm}^{-3}$ .
- Gas turbines have a maximum tar tolerance much lower; no more than  $0.5 \text{ mg.Nm}^{-3}$ . Concerning the solid particles maximum rate in the gas turbines, we didn't have confirmed

data but we can assume that it is in the same order of magnitude that for the Fischer Tropsch process.

- The Fischer Tropsch process requires tar concentrations below  $0.1 \text{ mg.Nm}^{-3}$  and a concentration in solid particles lower than  $0.02 \text{ mg.Nm}^{-3}$ .

In the major part of these applications, an external treatment of the crude syngas is mandatory in the aim to have a syngas as pure as possible.

### ***Tar destruction***

Tar condensing at low temperature, their presence in the crude gas is a technological problem for the gasifiers. However, this problem is not significant, as such all tars are at a temperature sufficient to be in gas phase but it is mandatory to process tars in function of the end-use.

The tar destruction can be divided into two methods: primary (inside the gasifier) and secondary (downstream of the gasifier) methods. The primary methods are more interesting because the thermodynamic efficiency losses associated to the gas cooling for its purification can be minimized. The ideal method is to use only the primary method [12].

### **Primary methods**

The primary methods are tar treatment inside the gasifier at high temperature. The main solutions proposed in the scientific literature are to optimize the design of the gasification reactor, its operating parameters (temperature, pressure, oxidizing agent/waste ratio, residence time ...), by adding catalyst or by plasma treatment [46-57]: The gasification temperature ( $> 1\,200 \text{ K} - 1\,300 \text{ K}$ ) has a beneficial effect to minimize the tar quantities and allows destroying the aromatics without a catalyst [47, 51]. A reduction of more than 40 % in tar yield has been reported when the temperature was raised from  $\sim 1\,000 \text{ K}$  to  $\sim 1\,200 \text{ K}$ . Preheating the gasification agent has the advantage to significantly reduce the tar content, due to the thermal decomposition at high temperature [50]. About the residence time, Kinoshita et al. [52] concluded that it has a little influence on the tar level, but it has significant influences on the tar composition. About the oxidizing agent/waste ratio, it is observed the decrease in tar yield by increasing this ratio [49]. Gasification can operate under atmospheric to high pressures. High pressures are preferred to directly connect the gas produced to downstream processes such as Fischer-Tropsch process (operating pressure around 30 bar), or gas turbines and synthesis of chemical products (up to 80 bar) [58]. To significantly reduce the energy

consumption in compression stages, gasifiers need to be sized at a pressure allowing the direct supply of syngas downstream units. It is preferable to compress the incoming agent such as reactant and product, than compress the gas output. According to [58], total energy spend in a coal gasification process is 22 MW at 5 bar and 5 MW at 50 bar for a gas production of  $100,000 \text{ Nm}^3 \cdot \text{h}^{-1}$ . The use of high pressure in the gasifier allows reducing the reactor size, to the detriment of their thickness, and heat losses in the compressor. However, increasing pressure favors the production of methane at low temperature [59], but becomes negligible at temperatures higher than 1 800 K.

Catalysts like dolomite, limestone, olivine sand, bauxite, lanthanum, alumina, nickel aluminate, cobalt, natural clay minerals and iron minerals can be used to optimize the tar reforming at high temperature [46, 53-57]. It is an efficient method for the tar destruction but this primary method can be very expensive in function of the catalyst used and its consumption.

### Secondary methods

Secondary methods suggest a cleaning downstream of the gasifier. Although the primary methods are the most important, the combination with downstream methods can be necessary in function of the tar destruction level requested. There are basically five systems of elimination of tars: thermal cracking, catalytic cracking, mechanical methods (scrubber, filter, cyclone, electrostatic precipitator ...), self-modifications (operating parameters) and plasma methods [12, 48].

Thermal and/or catalytic cracking is preferably used when the temperature of the gasifier does not allow the processing of tars by the primary methods ( $T < 1\,200 \text{ K}$ ). The cracking of tars by cold plasma in series with gasification reactor has been studied and several groups have demonstrated with success that organic elements (tars and particles) can be easily decomposed by corona discharges or by gliding-arc [48, 60-63]. The role of the plasma treatment is twofold: it allows, on the one hand, a significant purification of gas by limiting the production of tars and on the other hand, producing a synthesis gas enriched in hydrogen (water-gas shift reaction). Plasma methods have also the advantages to be able to operate at high temperature and to be retrofitted to existing installation.

The tar removal by secondary methods is one of the most concern topics for current scientific research and numerous treatment methods regularly emerge from the scientific community

and are reported to be very effective in tar reduction but still need to be optimized to be economically viable and used industrially [46, 47].

Rabou et al. [51] preconize a residence time of 4s for the thermally conversion of 97 % of tar in a dense fluidized bed at 1 450 K. Fourcault et al. [64] demonstrate the influence of plasma parameters on the tar destruction with a small influence on the concentration of synthesis gas. Gasification by thermal plasma gives good results in terms of gas purity and energy efficiency. The main difference from conventional methods comes from the tar concentration at the output of these processes. Thus, the autothermal methods provide tar content exceeding  $1000 \text{ mg.m}^{-3}$  while it is on the order of  $1 \text{ mg.m}^{-3}$  for plasma processes. The tar contents at the output of a gasification stage by thermal plasma are 1000 time less than that obtained by autothermal fluidized bed [46]. These results can be very interesting for an application of synthesis gas in second generation biofuel that requires tar concentration below  $0.1 \text{ mg.m}^{-3}$ . However, to reach this threshold concentration, the purification of syngas is mandatory but will be less costly. According to Göransson et al. [12], the drying of the hot gas under high pressure can provide a highly effective removal of contaminants. This technique still requires the use of high pressures to reduce heat losses in the compressor.

### **Thermal plasma gasification processes**

Allothermal gasification processes requires external energy source which can be of different nature: external pre-heating of the reactor by combustion, electrical energy, solar energy...

Considering the performances, it appears from literature than plasma seems to be one of the most probative technologies for waste gasification. In this review, only the allothermal gasification processes based on high temperature plasma have been studied.

However, plasmas technologies applications are not recent. Plasmas processes have been used and developed during the nineteenth century by the metalworking industry to provide extremely high temperatures in furnaces. During the early twentieth century, plasma processes were used in the chemical industry to manufacture acetylene from natural gas. Since early 80's, plasma technology is considered as a highly attractive route for the processing of MSW and successful applications in treatment of hazardous and harmful materials such as asbestos, radioactive waste vitrification and chemicals have showed the maturity of this technology [23-29, 65-70].

### **Advantages of the plasma technology**

Knoef [34] shows the differences obtained in autothermal processes in function of the oxidant used (pure oxygen or air). Pure oxygen provides a gas with a low calorific value of  $10.1 \text{ MJ.m}^{-3}$  while the use of air gives only  $4.2 \text{ MJ.m}^{-3}$  due to the dilution of the synthesis gas with the nitrogen introduced with the air flow. Although, using a high flow rate of pure oxygen is expensive for an industrial process. Steam water is generally preferred because it produces the desired reactions including the steam reforming reaction and increases the  $\text{H}_2$  ratio in the syngas. However, the steam reforming reaction is highly endothermic and need high temperature (1100 K-1700 K). One solution is to use a dual fluidized bed reactor. This reactor is designed to separate the gasification from the combustion. The high temperatures are conveyed between the two reactors with sand. The interest of this technology is to prevent syngas dilution by the nitrogen of the air and the combustion of the wastes by separating the gasification from the combustion. The high temperature obtained in the reactor without using the combustion process allows producing a synthesis gas with high purity and high calorific value.

Autothermal processes involve chemical reactions known as redox. These reactions establish high temperatures in the reactor but the maximum temperature possible by combustion is 3000K [31] (for acetylene-oxygen mixture) while it is possible to achieve gas temperature up to 15 000K with a thermal plasma [71]. Such a temperature in plasmas can allow synthesizing or degrading chemical species in some conditions unreachable by conventional combustion and can greatly accelerate the chemical reactions. Thermochemistry of combustion does not allow precise control of the enthalpy injected into the reactor. Plasma process allows an easiest enthalpy control by adjusting the electrical power. The reactive species produced by the plasma, such as atomic oxygen and hydrogen or hydroxyl radicals, is an additional advantage for the use of plasma. In the literature, it is reported that these species enhance strongly the degradation of the tars with greater efficiency than conventional processes [72, 73].

Only dual fluidized bed or allothermal processes can allow producing syngas without nitrogen dilution. Nevertheless, dual fluidized bed produces high tar content, mostly above  $10 \text{ g.Nm}^{-3}$  [12].

These technical limitations lead to consider the plasma technology as one of the best alternative approaches to produce high purity and high LHV syngas.

### **Plasma torch configurations in the reactor**

In a first configuration, the product is injected close to the plasma plume. This method is the most studied in the literature with scientific experiments in China [25, 73-76], Taiwan [77], France [78], Czech Republic [71, 79-84] and Russia [85-87]. The product can be injected in solid form [71] or liquid [78], from a pyrolysis cycle or crushed. The torch is often located in the upper part of the reactor with an ash recovery area in the bottom part [32, 81].

In a second configuration, plasma torches are close to a molten bath. Indeed, only inorganic compounds form a melt. However, these technologies are mainly used for treatment of industrial wastes like asbestos or radioactive waste [88]. Steam is injected in the treatment zone to obtain gasification reaction. At the reactor output, other steam injection allows to stop the chemical reactions by quenching. Moreover, steam helps to adjust the  $H_2/CO$  ratio by the water gas shift reaction.

A final thermal plasma configuration process is a hybrid process, incorporating plasma technology combined with incineration or some other thermal processing technology, usually allows better use of the heating value of the waste material. This configuration, which cannot be considered purely as a thermal plasma gasification technology, is a thermal plasma treatment of gases leaving the reactor. This treatment technique is used in the plants of CHO power, Plasco Energy group, AlterNRJ. In this case, thermal plasma allows the tar treatment of the syngas at the exit of the gasification reactor [64]. The pyrolysis zone (autothermal zone) is separated from the reduction zone (plasma zone). This configuration is similar to a two-stage gasifier design which is reported to be very effective in producing clean gas [41]. In [89], they concluded that the tar content was 40 times less with a two-stage gasifier (about  $50 \text{ mg.m}^{-3}$ ) than with a single-stage reactor under similar operating conditions.

### **Waste gasification by plasma process at lab scale**

In this part of the review are compared the performances of various waste gasification processes based on thermal plasma represented in the scientific literature. Thermal plasmas can be obtained by arc discharges (DC or AC plasma) or by Radio Frequency (RF)-MicroWave (MW) plasma. RF or MW plasmas don't have electrode erosion but have low energy efficiency, about 40 % to 70 % (60 % to 90 % for arc torches) [73]. From this review, it clearly appears that the main plasma torch technology applied to waste gasification is the DC technology which is used in all the studies excepted in [32, 85, 90] where an AC plasma torch is used.

A wide range of organic wastes treated by plasma process has been studied: MSW [9, 23, 66, 91-94], used tires [24, 25, 73, 75, 85], paper mill waste [95], plastic waste [73, 74, 96, 97], liquid and solid hazardous waste [98-101], Refuse Derived Fuel (RDF, i.e. mixture of plastics, paper, wood and dried organic material) [67, 85, 86], medical Waste [66, 73] and biomass wastes [32, 66, 71, 73, 76, 77, 81, 85, 90, 102-104].

In [66], one can notice that the  $H_2$  and CO yields are strongly linked to the original composition of the waste treated and can vary, with the same experimental conditions, from 49.4 vol% to 64.4 vol% for  $H_2$  and from 24.8 vol% to 36 vol% for CO in function of the nature of the waste. It is important to keep in mind that the  $H_2$  and CO rates in the crude syngas is strongly linked to the oxidizing agent injected and/or the moisture content in the waste treated. In most of the cases, the results given by authors don't take into account the nitrogen and the water included in the crude syngas. Moreover, the temperature of exhaust gas, the pressure, the waste flow rate injected, the oxidizing agents such as steam or air are rarely expressed with precision. All these lacks of information make it difficult to accurately determine the influence of the plasma on the syngas product.

Based on this review, it cannot be concluded on the best effective configuration from the different experimental results which are strongly linked to the reactor used and the elementary composition of the waste treated. However, the performance and composition of synthesis gas differs significantly depending on the method applied and gasification parameters as follows:

- the elementary composition of the waste (carbon, hydrogen, oxygen, moisture, mineral elements)
- the LHV of the waste
- the nature of the oxidizing agent (air,  $O_2$ ,  $CO_2$ , steam)
- the amount of injected oxidant (water gas shift reaction increases the rate of hydrogen)
- reactor pressure
- the temperature gradient within the reactor
- the scale effect and heat losses as a function of the refractory material of the reactor
- the quality of the post-treatment of the crude syngas
- the influence of pretreatment of the product before the injection
- the mix of the waste with a fossil fuel (coal powder) to optimize the temperature and the reactions

On plasma technology must be added the nature of the plasma gas (Ar,  $N_2$ ,  $H_2O$ ,  $H_2$ , CO,  $CO_2$ ...), the specific enthalpy, the diffusion rate of plasma, the injected power, the thermal efficiency of the plasma torch and the technology of the plasma torch (DC, AC or RF).

Considering relevant performances, it appears clearly from literature that plasma appears to be one of the most probative technologies for waste gasification. Determinant performances are the CO and H<sub>2</sub> yields, the cold gas efficiency (energy efficiency), the gas calorific value (LHV) and particularly, the tar content. This last information is often neglected by the authors and is only available in specific studies based on the post-treatment of the tar in a second step of the processes by primary or secondary methods.

Due to the absence of complete results in the scientific literature, only few representative plasma gasification processes have been compared [25, 32, 74, 81, 85, 92, 94].

In Table 3 are summarized the main results on plasma gasification processes and are compared to typical ranges of variations of some operating and process performance parameters in autothermal gasification of MSW [9].

These results give tendencies about the efficiency of allothermal processes. The main conclusion is that allothermal gasification (DC or AC torch) allows processing all kind of wastes by adjusting the energy input with the plasma, independently of the oxidizing agent ratio, the LHV and the moisture of the waste. The LHV of the cold gas is function of the nature of the waste treated and the nature of the oxidizing agent but for the different studies, independently of these parameters and the plasma technology used, the energy efficiencies are in the same order of magnitude and are comparable to the typical range of energy efficiency in autothermal gasification.

About the comparison of allothermal versus autothermal gasification processes, the main difference is on the net electrical efficiency (15 % - 24 % for autothermal and 26 % - 49 % for allothermal) which is based on the theoretical electrical conversion performances of the end-use devices which are strongly dependent of the tar content in the syngas, key parameter for the performances of the overall process.

The high enthalpy, the residence time and high temperature in plasma can advantageously improve the physical conditions for gasification, which are inaccessible in other thermal processes and can enhance strongly the degradation of the tars and allow reaching, due to low tar content in the syngas, better net electrical efficiency than autothermal processes.

Best performances have been obtained by Rutberg and al. [32] who have studied, experimentally (plasma process) and numerically (autothermal versus plasma), the gasification of wood residues with different oxidizing agents (Air, O<sub>2</sub>, CO<sub>2</sub>, H<sub>2</sub>O and mixture of these oxidizing agents). The results show the benefit of the allothermal plasma process versus the autothermal process with the adding of an oxidizing agent. Experimentally, the best results are that from 1 kg of wood residues with a moisture of ~ 20 % (LHV = 13.9 MJ.kg<sup>-1</sup>),



it is possible, by plasma gasification with air as oxidizing agent, to generate  $\sim 13.5$  MJ of chemical energy with an energy consumption of  $\sim 2.16 \text{ MJ.kg}^{-1}$  (plasma) and allows to generate  $\sim 8.58 \text{ MJ.kg}^{-1}$  of electric energy. The energy efficiency is  $\sim 84 \%$  and the net electrical efficiency is  $\sim 46.2 \%$ . This net electrical efficiency is based on the fact that the energy consumption of the plasma ( $2.16 \text{ MJ.kg}^{-1}$ ) comes from the  $8.58 \text{ MJ.kg}^{-1}$  of electric energy generated by the process  $((8.58 - 2.16) / 13.9 = 0.462)$ . These results have been obtained experimentally with a lab-scale 3 Phase AC plasma torch whose the characteristics have been fully detailed in numerous papers [32, 85, 86, 105-112]. In their theoretical study, they indicate that the incorporation of  $\text{CO}_2$  or  $\text{H}_2\text{O}$  in the plasma gasification of wood residues has the advantage to perfectly control the  $\text{H}_2/\text{CO}$  ratio in the output gas.

### **Overview of waste gasification by plasma process at industrial scale**

The current market for high power plasma torches is mainly shared by four companies: Westinghouse, Europlasma, Tetronics and Phoenix Solutions Company (PSC). The technologies developed by Westinghouse [8, 113, 114], Europlasma [78, 92, 115] and PSC [116] are based on transferred and non-transferred DC torches with water-cooled metal electrodes while Tetronics torch [117, 118] is based on a transferred DC torch with two graphite electrodes not water cooled. Advanced Plasma Power (APP) and Tetronics have a collaboration agreement for the development and commercialization of plasma gasification WTE plants based on the technology of transferred DC torch [119]. For Westinghouse and Europlasma, their strategy is different since they have each developed a plasma gasification WTE process based on their own DC torch technology and market turnkey plants through subsidiaries (Alter NRG for Westinghouse and CHO-Power for Europlasma, respectively) [120, 121]. In parallel of these developments of industrial plasma gasification WTE plants, some companies also develop their own facility based on Westinghouse, Europlasma or PSC DC Torches (such as Plasma Arc Technologies, Plasco Energy Group, Enersol Technologies, Bellwether Gasification Technologies, Startech Environmental, Green Power Systems, Hitachi Metals ...) [122-127] or on home-made torches (PEAT, InEnTec, Pyrogenesis ...) [128-130]. Often, there is very few information on home-made torch technologies developed but it seems to be mainly based on DC torches. Although not yet validate for the waste gasification at an industrial scale, other plasma torch technologies (RF and AC) are being developed at a pilot scale in several research laboratories such as Applied Plasma

Technologies (USA), PERSÉE - MINES ParisTech (France) or Institute for Electrophysics and Electric Power - Russian academy of sciences (Russia).

Among the various gasification waste-to-energy processes at industrial scale, it follows two main configurations for the location of the plasma torches in the reactor which are mainly based on the waste to be treated. For the gasification of waste with low organic matter content, it is necessary to treat the waste at high temperature in order to melt the inorganic part. The products obtained are syngas from the organic part of the waste and slag from the non-organic part of the waste. In this case, the plasma torches are placed in the reactor body closest to the molten bath and the torches are non-transferred arc or transferred arc (the bath playing the role of anode for the plasma torch). In the case of waste with a high proportion of organic matter, it is not necessary to raise the temperature of the reactor above 1 800 K and in this case, the waste gasification in the reactor can be made either by autothermal or allothermal ways (plasma, dual fluidized beds ...). In this case, the plasma torch is placed at the outlet of the gasification reactor before the cooling of the crude syngas in the aim to treat the tar content in the syngas at an optimized energy cost (primary method).

In the last part of this study are detailed the various technologies of high power plasma torches, their levels of development, their gasification efficiency and the current status of waste plasma gasification plants in the world.

### ***DC Torches***

Westinghouse [114] is an American company created in the 1970s. The first R & D on the application of plasma began in partnership with NASA on the development of plasma torches to recreate and simulate the entry conditions in the atmosphere of the probes of the space program “Apollo”. It is only in the 1980s that the use of the torches has evolved to the waste treatment at high temperature. Currently, the company markets several torches, offering a wide range of power from 5 kW to 2400 kW. In the Table 4 are described the main characteristics of the Westinghouse DC torches [114]. These torches generally operate in non-transferred arc and can use different plasma gases: air, oxygen, nitrogen...

Europlasma is a French company created in the 90s by EADS-LV (Formerly Aerospatiale) [115]. Like Westinghouse, this technology was originally developed for space and military applications before developing applications related to the steel industry and the recovery and waste treatment. Today, the company markets a wide range of DC plasma torches (Table 5) whose powers range from 80 kW to 4000 kW depending on the type of application (gas

treatment, waste and biomass gasification (CHO-Power, subsidiary of Europlasma) [121], asbestos destruction (INERTAM, subsidiary of Europlasma) [131].

DC Plasma torches developed by Europlasma are very similar to Westinghouse torches.

A peculiarity of the Europlasma system is linked to the upstream electrode which is surrounded by a coil which generates a magnetic field. This allows controlling the movement of the extremity of the arc at the upstream electrode. At the downstream electrode, the arc movement is governed by the gas flowing into the injection chamber. The electrodes are cooled by a deionized water pressurized system.

Phoenix Solutions Company (PSC) is an American company created in 1993 (formerly Fluidyne Engineering Corp.) [116]. Fluidyne, after 40 years at the forefront of aeronautical and aerospace researches, was reformed as PSC in 1993 and became, in 20 years, one of the largest suppliers of plasma heating systems in the world. Today, the company markets a wide range of DC plasma torches (Table 6) whose powers range from 50 kW to 3000 kW and can operate on different plasma gas (air, N<sub>2</sub>, O<sub>2</sub>, H<sub>2</sub>, CO or CO<sub>2</sub>). Their plasma torches are used in several plasma gasification WTE plants [8, 124-127].

The Tetronics plasma torches are based on DC technologies which can be transferred arc or non-transferred arc, with graphite electrodes or based on the TwinTorch™ system, wherein, two transferred arc torches are of opposite polarity connected in series [117, 118]. These different torches represent the basis of all the different waste treatment devices proposed by Tetronics. The advantage of TwinTorch™ system is the electrodes in graphite which can be adjusted in function of their erosion. However, the investment cost for this technology is expensive due to the use of a DC power supply (fragile technology with frequent maintenance). Tetronics is the plasma torch supplier of Advanced Plasma Power which has several projects of plasma gasification of MSW in UK.

For around 10 years, 100 kW DC plasma torches are widely used in China for pulverized-coal boiler burners (Guodian plasma torches). Their technology is very similar to the Messerle first-generation plasma ignition system [132]. Oral statements of industrial representatives claim that over 400 systems are in operation in China, but not for gasification, mainly due to the limited lifetime of the electrode (less than 100 hours).

### ***RF Torches***

Applied Plasma Technologies (APT) is working on the development of high power hybrid RF + DC plasma torches [133-141]. Like many researchers, it appears to them that a major

shortcoming of DC plasma torches is the short lifetime of the cathode and think RF technology is the solution to solve this major problem. Currently, they have developed a hybrid plasma torch (RF + DC) that has good energy efficiency (between 80 % and 95 %) for a power of 150 kW [140]. It is sure that this technology has, on the one hand, the great advantage of not having electrode erosion, but, on the other hand, has a very limited power with high investment costs. APT is currently working on developing a 1 MW torch but is still faced with power limitation and the expensive cost of the RF technology.

### ***AC Torches***

Some non-transferred 3-phase AC plasma torches used for waste, biomass or coal processing have been described by the IEEP team (Institute for Electrophysics and Electric Power - Russian academy of sciences) in several papers [32, 85, 86, 105-112]. The electrodes consist of water-cooled copper tubes, and the movement of the arc attachment using the self-magnetic field of the current in the electrodes (rail gun effect) minimizes erosion. This plasma torch has been designed to work on oxidizing media. A new AC plasma torch for gasification has recently been developed and can work with steam as plasma gas. The plasma torch is able to work stationary on air with electrical power from 100 kW to 600 kW. Electrode erosion is again the weakness of this technology whose the electrodes lifetime is limited to 200 hours.

An original semi-industrial scale plasma technology using a three-phase AC source is presently working at the “Centre Procédés, Energies Renouvelables et Systèmes Energétiques” – PERSÉE – MINES ParisTech in Sophia-Antipolis, France. This technology has been developed initially for the synthesis of carbon nanoparticles like fullerenes, carbon blacks, nanotubes and others. This technology has evolved since 1993 and has reached a high level of reliability, unique at this scale [142, 143].

The operating principle is as follows: plasma gas is introduced through the upper part of the plasma torches and surrounding of the graphite electrodes. An electrical arc is initiated between the three graphite electrodes, each electrode being supplied by one phase of the three-phase AC power supply. The main characteristics of the power supply are summarized in Table 7. The electrodes, alternatively anode and cathode, are the points of attachment of the arcs. The plasma generated consists of free arcs rotating with the frequency of the current.

The 3-phase AC plasma Torch (TAT) has some similarities with electrometallurgy technologies, e.g. electric arc furnaces for steelmaking and submerged arc furnaces for silicon

metal and ferrosilicon production. It allows large high temperature volumes and long residence times so, it is particularly adapted to the waste gasification.

Contrary to three-phase arcs commonly found in metallurgical applications, there is not a neutral point in this developed system. Thus, the arcs move freely interacting mutually by electromagnetic forces. This electromagnetic interaction leads to a global motion in the centrifugal direction [144, 145].

In the case of the gasification process, it is necessary to protect the electrodes which are in graphite, from the oxidative medium. The graphite electrode erosion can be minimized by using a sheathing gas.

Some companies are working on plasma gasification of waste such as InEnTec or PEAT but no communication is done on the technology of the plasma torches used in their plasma processes marketed as well as on their power. All their communications are based on the gasification process performances in terms of quantity of waste processed and output electrical power. In general, the gasification processes proposed are mainly based on the use of plasma torches for the vitrification of solid residues obtained after waste gasification and for the refining of the crude syngas obtained after waste gasification in the aim to have a high purity syngas, necessary for their end-use in high conversion energy processes such as gas turbines or fuel cells, or for the production of synthesis fuel.

The current market for waste plasma gasification plants is shared by some companies including Alter NRG (Westinghouse subsidiary), CHO-Power (Europlasma subsidiary), Advanced Plasma Power (APP), InEnTec, Hitachi Metals Ltd., Plasco Energy Group Inc., EnviroParks Limited [146], Sunbay Energy Corporation [147], Green Power Systems, Pyrogenesis [94, 130, 148-150], PEAT...

In Tables 8 and 9 are listed the main plants for waste gasification by plasma currently in operation around the world and the numerous plant projects for APP, Alter NRG and CHO Power. Presently, the technical feasibility and economical viability of plasma vitrification technologies have been demonstrated for a large range of hazardous wastes but it is not totally the case of plasma gasification technologies for the disposal of MSW at an industrial scale. This is a growing market and the efficiency of the waste gasification by plasma seems to be validated but the economic viability of this technology must be proven before to be accepted by the industry [4, 68, 151-160].

However, plants currently installed are of medium size (few MW of electricity produced by plant). They are mainly demonstration units to promote plasma technology for waste gasification. The proposed technologies appear fully functional but it seems that the limit of

installed plants is mainly due to power limitations of plasma torches and DC power supplies available on the market (around one MW) and their high investment cost and maintenance cost. Presently, most of the gasification technologies are based on DC plasma torches. Although widely used, these technologies have strong technical and economic constraints mainly related to their lack of robustness and reliability and their equipment and operating costs relatively high due to their frequent and expensive maintenance: (i) short lifetime of the electrodes (between 300 h and 500 h ) [98], (ii) sensitive electronics. Indeed, the technological complexity of the power supplies of the DC torches involves a costly price, mainly due to the rectifier part of the electrical signal that can involve an increase of 30 % of the price of the power supply. Moreover, the limited autonomy of the metal electrodes implies high operating costs. AC power supplies could be an alternative for reducing costs. They have proven their durability and reliability for many years in the steel industry with investment cost and maintenance cost much cheaper for few MW. For the future development at industrial scale of the Waste-to-Energy gasification processes based on thermal plasma, it appears that it will be mandatory to overcome these limitations of robustness as well as to significantly reduce equipment and operating costs. Plasma torches based on cheap consumable parts like graphite electrodes can avoid their water cooling, making this plasma technology less complex and more reliable and could be a solution to the problems of reliability and equipment/operating costs for the development of the plasma gasification at industrial scale.

From the perspective of life cycle assessment of three different technologies of thermal plasma generator (Radio-frequency plasma system – RF, microwave-induced system – MW and plasma torch system – PT) in comparison of downdraft gasifier system – DG, a recent comparative study of Shie et al. [160] on the overall thermal efficiency –  $\eta_E$ , the energy return on investment – EROI and the net energy ratio – NER, indicate the highest efficiency of the PT technology in all the cases with a  $\eta_E$  of 84.07 % (64.11 % for DG, 38.59 % for MW and 57.03 % for RF), with a NER of 7.86 (5.79 for DG, 3.13 for MW and 5.01 for RF) and an EROI of 8.86 (6.79 for DG, 4.13 for MW and 6.01 for RF). These terms, according to the authors, are more representative than the price of production which is influenced by the markets.

## **Conclusion**

In this review, where are compared the performances of the different waste gasification processes based on thermal plasma represented in the scientific literature, the main

conclusions are that plasma technology appears to be one of the most probative technologies for the processing of waste-to-energy and can be easily adapted to the treatment of various wastes (municipal solid wastes, heavy oil, used car tires, medical wastes ...). Allothermal gasification allows processing all kind of wastes (domiciliary and industrial wastes) by adjusting the energy input with the plasma, independently of the oxidizing agent ratio, the LHV and the moisture of the waste.

About the comparison of chemical efficiency of allothermal versus autothermal gasification processes: (i) the values of cold gas efficiency (energy efficiency) are in the same order of magnitude, (ii) the main difference is on the net electrical efficiency of the overall process. This value is based on the theoretical electrical conversion performances of the end-use devices which are strongly dependent of the tar content in the syngas, key parameter for the performances of the overall process. The high enthalpy, the residence time and high temperature in plasma can advantageously improve the conditions for gasification, which are inaccessible in other thermal processes and can enhance strongly the degradation of the tars and allow reaching, due to low tar content in the syngas, better net electrical efficiency than autothermal processes.

Concerning the advantages of the waste gasification by thermal plasma, the role of the plasma treatment is twofold: it allows, on the one hand, a significant purification of gas by limiting the production of tars and on the other hand, producing a synthesis gas enriched in hydrogen (water-gas shift reaction). Plasma methods have also the advantages to be able to operate at high temperature and to be retrofitted to existing installation. Such a temperature in plasmas can allow synthesizing or degrading chemical species in some conditions unreachable by conventional combustion and can greatly accelerate the chemical reactions. Thermochemistry of combustion does not allow precise control of the enthalpy injected into the reactor. Plasma process allows an easiest enthalpy control by adjusting the electrical power. The reactive species produced by the plasma, such as atomic oxygen and hydrogen or hydroxyl radicals, is an additional advantage for the use of plasma and enhance strongly the degradation of the tars with greater efficiency than conventional processes.

Concerning the development and the operation of the plasma technologies on the energy market, presently, the technical feasibility and economical viability of plasma vitrification technologies have been demonstrated for a large range of hazardous wastes but it is not totally the case of plasma gasification technologies for the disposal of MSW at an industrial scale. This is a growing market and the efficiency of the waste gasification by plasma seems to be validated but the economic viability of this technology must be proven before to be accepted

by the industry. Presently, the strong expansion in the world of numerous plasma gasification plants (projects and operational plants) shows clearly that a step has been taken and in the future, plasma gasification will play a significant role in the field of renewable energy.

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## References

1. Nishikawa, H., Ibe, M., Tanaka, M., Takemoto, T., Ushio, M.: Effect of DC steam plasma on gasifying carbonized waste. *Vacuum* **80**, 1311-1315 (2006)
2. Nzihou, A.: Toward the valorization of waste and biomass. *Waste & Biomass Valor.* **1**, 3-7 (2010)
3. Zhang, L., Xu, C., Champagne, P.: Overview of recent advances in thermo-chemical conversion of biomass. *Energy Convers. and Manag.* **51**, 969-982 (2010)
4. Bosmans, A., Vanderreydt, I., Geysen, D., Helsen, L.: The crucial role of Waste-to-Energy technologies in enhanced landfill mining: a technology review. *J. of Clean. Prod.* – Article in press – doi:10.1016/j.jclepro.2012.05.032 (2012)
5. Di Gregorio, F., Zaccariello, L.: Fluidized bed gasification of a packaging derived fuel: energetic, environmental and economic performances comparison for waste-to-energy plants. *Energy* **42**, 331-341 (2012)
6. Castaldi, MJ., Themelis, NJ.: The case for increasing the global capacity for waste to energy (WTE). *Waste & Biomass Valor.* **1**, 91-105 (2010)
7. Yang, H., Yan, R., Chen, H., Lee, DH., Liang, DT., Zheng, C.: Pyrolysis of palm oil wastes for enhanced production of hydrogen rich gases. *Fuel Process. Technol.* **87**, 935-942 (2006)
8. Review of technologies for gasification of biomass and wastes. E4tech June 2009, <http://www.nnfcc.co.uk/tools/review-of-technologies-for-gasification-of-biomass-and-wastes-nnfcc-09-008> (2009)
9. Arena, U.: Process and technological aspects of municipal solid waste gasification. A review. *Waste Manag.* **32**, 625-639 (2012)
10. Ricketts, B., Hotchkiss, R., Livingston, B., Hall, M.: Technology status review of waste/biomass co-gasification with coal. IChemE Fifth European Gasification Conference, 8-10 April 2002, Noordwijk, The Netherlands (2002)
11. Morris, M., Waldheim, L.: Energy recovery from solid waste fuels using advanced gasification technology. *Waste Manag.* **18**, 557-564 (1998)
12. Göransson, K., Söderlind, U., He, J., Zhang, W.: Review of syngas production via biomass DFBGs. *Renew. and Sustainable Energ. Reviews* **15**, 482-492 (2011)
13. Manyà, JJ., Sánchez, JL., Ábrego, J., Gonzalo, A., Arauzo, J.: Influence of gas residence time and air ratio on the air gasification of dried sewage sludge in a bubbling fluidised bed. *Fuel* **85**, 2027-2033 (2006)



14. Kim, HY.: A low cost production of hydrogen from carbonaceous wastes. *International J. of Hydrogen Energ.* **28**, 1179-1186 (2003)
15. Zhao, Y., Sun, S., Zhou, H., Sun, R., Tian, H., Luan, J., Qian, J.: Experimental study on sawdust air gasification in an entrained-flow reactor. *Fuel Process. Technol.* **91**, 910-914 (2010)
16. Yassin, L., Lettieri, P., Simons, S., Germanà, A.: Techno-economic performance of energy-from-waste fluidized bed combustion and gasification processes in the UK context. *Chem. Eng. J.* **146**, 315-327 (2009)
17. Arena, U., Di Gregorio, F., Amorese, C., Mastellone, ML.: A techno-economic comparison of fluidized bed gasification of two mixed plastic wastes. *Waste Manag.* **31**, 1494-1504 (2011)
18. Xiao, G., Ni, MJ., Chi, Y., Jin, BS., Xiao, R., Zhong, ZP., Huang, YJ.: Gasification characteristics of MSW and an ANN prediction model. *Waste Manag.* **29**, 240-244 (2009)
19. Arena, U., Zaccariello, L., Mastellone, ML.: Fluidized bed gasification of waste-derived fuels. *Waste Manag.* **30**, 1212-1219 (2010)
20. Surisetty, VR., Kozinski, J., Dalai, AK.: Biomass, availability in Canada, and gasification: an overview. *Biomass Conv. Bioref.* **2**, 73-85 (2012)
21. Belgiorno, V., De Feo, G., Della Rocca, C., Napoli, RMA.: Energy from gasification of solid wastes. *Waste Manag.* **23**, 1-15 (2003)
22. Faaij, A., Van Ree, R., Waldheim, L., Olsson, E., Oudhuis, A., Van Wijk, A., Daey-Ouwens, C., Turkenburg, W.: Gasification of biomass wastes and residues for electricity production. *Biomass and Bioenergy* **12**(6), 387-407 (1997)
23. Cherednichenko, VS., Anshakov, AS., Danilenko, AA., Michajlov, VE., Faleev, VA., Kezevich, DD.: Domestic waste plasma gasification technology and its comparison with ordinary one burning on the final products. *Ecology, Electrotechnology and Waste Process. – KORUS 2002*. 211-213 (2002)
24. Chang, JS., Gu, BW., Looy, PC., Chu, FY., Simpson, CJ.: Thermal plasma pyrolysis of used old tires for production of syngas. *J. of Environ. Sci. and Health. Part A* **31**(7), 1781-1799 (1996)
25. Huang, H., Tang, L., Wu, CZ.: Characterization of gaseous and solid product from thermal plasma pyrolysis of waste rubber. *Environ. Sci. Technol.* **37**, 4463-4467 (2003)
26. Cheng, TW., Chu, JP., Tzeng, CC., Chen, YS.: Treatment and recycling of incinerated ash using thermal plasma technology. *Waste Manag.* **22**, 485-490 (2002)
27. Chu, JP., Hwang, IJ., Tzeng, CC., Kuo, YY., Yu, YJ.: Characterization of vitrified slag from mixed medical waste surrogates treated by a thermal plasma system. *J. of Hazard. Mater.* **58**, 179-194 (1998)
28. Tzeng, CC., Kuo, YY., Huang, TF., Lin, DL., Yu, YJ.: Treatment of radioactive wastes by plasma incineration and vitrification for final disposal. *J. of Hazard. Mater* **58**, 207-220 (1998)
29. Mohai, I., Szépvölgyi, J.: Treatment of particulate metallurgical wastes in thermal plasmas. *Chem. Eng. and Process.* **44**, 225-229 (2005)
30. Colby, S., Turner, Z., Utley, D., Duy, C.: Treatment of radioactive reactive mixed waste. *WM'06 Conference*, February 26-March 2, Tucson, AZ (2006)
31. Fauchais, P.: Technologies plasma : applications au traitement des déchets. *Techniques de l'Ingénieur G2055*:1-11 (2007)
32. Rutberg, PhG., Bratsev, AN., Kuznetsov, VA., Popov, VE., Ufimtsev, AA., Shtengel', SV.: On efficiency of plasma gasification of wood residues. *Biomass and Bioenergy* **35**, 495-504 (2011)

33. Boulos, M.I.: New frontiers in thermal plasma processing. *Pure & Appl. Chem.* **68**(5), 1007-1010 (1996)
34. Knoef, HAM. Handbook. Biomass Gasification. BTG, Netherlands, ISBN: 90-810068-2005-380. (2005)
35. Khodakov, A.Y., Chu, W., Fongarland, P.: Advances in the development of novel cobalt Fischer–Tropsch catalysts for synthesis of long-chain hydrocarbons and clean fuels. *ChemInform*, doi: 10.1002/chin.200733255 (2007)
36. Ladwig, M., Lindvall, K., Conzelman, R.: The realised gas turbine process with sequential combustion: Experience, state of development, prospects. *VGB Powertech* **87**(10), 30-35 (2007)
37. Ishikawa, M., Terauchi, M., Komori, T., Yasuraoka, J.: Development of high efficiency gas turbine combined cycle power plant. Mitsubishi Heavy Industries, Ltd. Technical review **45**(1), 15-17 (2008)
38. Galeno, G., Minutillo, M., Perna, A.: From waste to electricity through integrated plasma gasification/fuel cell (IPGFC) system. *Intern. J. of Hydrogen Energ.* **36**, 1692-1701 (2011)
39. Oxtoby, D. W.: Principles of Modern Chemistry (5th ed.). Thomson Brooks/Cole. ISBN 0-03-035373-4 (2002)
40. Bridgwater, A.V.: Renewable fuels and chemicals by thermal processing of biomass. *Chem. Eng. J.* **91**, 87-102 (2003)
41. Chopra, S., Jain, A.K.: A review of fixed bed gasification systems for biomass. *Agricultural Engineering International: the CIGR Ejournal*. Invited Overview 5:IX April 2007 (2007)
42. Reed, T.B., Gaur, S.: A survey of biomass gasification ISBN 1-890607-13-4180. (2001)
43. Petitpas, G., Rollier, J.D., Darmon, A., Gonzalez-Aguilar, J., Metkemeijer, R., Fulcheri, L.: A comparative study of non-thermal plasma assisted reforming technologies. *Intern. J. of Hydrogen Energ.* **32**:2848-2867 (2007)
44. Brage, C., Qizhuang, U., Sjostrom, K.: Use of amino phase adsorbent for biomass tar sampling and separation. *Fuel* **76**(2), 137-142 (1997)
45. Neeft, J.P.A., Knoef, HAM., Onaji, P.: Behaviour of tar in biomass gasification systems. Tar related problems and their solutions. November Report No. 9919. Energy from Waste and Biomass (EWAB), The Netherlands (1999)
46. Milne, T.A., Evans, R.J.: Biomass Gasifier Tars: Their Nature, Formation, and Conversions. National Renewable Energy Laboratory, Golden, CO. NREL/TP-570-25357 (1998)
47. Devi, L., Ptasiński, K.J., Janssen, F.: A review of the primary measures for tar elimination in biomass gasification processes. *Biomass and Bioenerg.* **24**, 125-140 (2003)
48. Han, J., Kim, H.: The reduction and control technology of tar during biomass gasification/pyrolysis: an overview. *Renew. and Sustainable Energ. Reviews* **12**, 397-416 (2008)
49. Mayerhofer, M., Mitsakis, P., Meng, X., De Jong, W., Spliethoff, H., Gaderer, M.: Influence of pressure, temperature and steam on tar and gas in allothermal fluidized bed gasification. *Fuel* **99**, 204-209 (2012)
50. Ponzio, A., Kalisz, S., Blasiak, W.: Effect of operating conditions on tar and gas composition in high temperature air/steam gasification (HTAG) of plastic containing waste. *Fuel Process. Technol.* **87**, 223-233 (2006)
51. Rabou, L., Zwart, R., Vreugdenhil, B., Bos, L.: Tar in biomass producer gas, the energy research centre of the Netherlands (ECN) experience: an enduring challenge. *Energ. Fuels* **23**, 6189-6198 – DOI:10.1021/ef9007032 (2009)

52. Kinoshita, CM., Wang, Y., Zhou, J.: Tar formation under different biomass gasification conditions. *J. of Analytical and Appl. Pyrolysis* **29**, 169-181 (1994)
53. Arena, U., Zaccariello, L., Mastellone, ML.: Tar removal during the fluidized bed gasification of plastic waste. *Waste Manag.* **29**, 783-791 (2009)
54. Graciosa Pereira, E., Nogueira Da Silva, J., De Oliveira, J., Machado, C.: Sustainable energy : a review of gasification technologies. *Renew. and Sustainable Energ. Reviews* **16**, 4753-4762 (2012)
55. Xu, C., Donald, J., Byambajav, E., Ohtsuka, Y.: Recent advances in catalysts for hot-gas removal of tar and NH<sub>3</sub> from biomass gasification. *Fuel* **89**, 1784-1795 (2010)
56. Nemanova, V., Nordgreen, T., Engvall, K., Sjöström, K.: Biomass gasification in an atmospheric fluidised bed: tar reduction with experimental iron-based granules from Höganäs AB, Sweden. *Catalysis Today* **176**, 253-257 (2011)
57. Bangala, D., Abatzoglou, N., Martin, JP., Chornet, E.: Catalytic gas conditioning: application to biomass and waste gasification. *Ind. Eng. Chem. Res.* **36**, 4184-4192 (1997)
58. Higman, C., Van Der Burgt, M.: *Gasification*. Oxford, S. 94, ISBN 978-0-7506-8528-3 (2008)
59. Valin, S., Ravel, S., Guillaudeau, J., Thiery, S.: Comprehensive study of the influence of total pressure on products yields in fluidized bed gasification of wood sawdust. *Fuel Process. Technol.* **91**, 1222-1228 (2010)
60. Nair, SA., Pemen, AJM., Yan, K., Van Heesch, EJM., Ptasiński, KJ., Drinkenburg, AAH.: Chemical processes in tar removal from biomass derived fuel gas by pulsed corona discharges. *Plasma Chem. and Plasma Process.* **23**(4), 665-680 (2003)
61. Nair, SA., Yan, K., Safitri, A., Pemen, AJM., Van Heesch, EJM., Ptasiński, KJ., Drinkenburg, AAH.: Streamer corona plasma for fuel gas cleaning: comparison of energization techniques. *J. of Electrostatics* **63**, 1105-1114 (2005)
62. Luche, J., Falcoz, Q., Bastien, T., Leninger, JP., Arabi, K., Aubry, O., Khacef, A., Cormier, JM., Lédé, J.: Plasma treatments and biomass gasification. *IOP Conf. Ser.: Mater. Sci. Eng.* **29** 012011 DOI:10.1088/1757-899X/29/1/012011 (2012)
63. Glocker, B., Nentwig, G., Messerschmid, E.: 1-40 kW steam respectively multi gas thermal plasma torch system. *Vacuum* **59**, 35-46 (2000)
64. Fourcalt, A., Marias, F., Michon, U.: Modelling of thermal removal of tars in a high temperature stage fed by a plasma torch. *Biomass and Bioenerg.* **34**, 1363-1374 (2010)
65. Cheng, TW., Tu, CC., Ko, MS., Ueng, TH.: Production of glass-ceramics from incinerator ash using lab-scale and pilot-scale thermal plasma systems. *Ceramics International* **37**, 2437-2444 (2011)
66. An'Shakov, AS., Faleev, VA., Danilenko, AA., Urbakh, EK., Urbakh, AE.: Investigation of plasma gasification of carbonaceous technogeneous wastes. *Thermophysics and Aeromechanics* **14**(4), 607-616 (2007)
67. Lemmens, B., Elslander, H., Vanderreydt, I., Peys, K., Diels, L., Oosterlinck, M., Joos, M.: Assessment of plasma gasification of high caloric waste streams. *Waste Manag.* **27**, 1562-1569 (2007)
68. Gomez, E., Amutha Rani, D., Cheesemen, CR., Deegan, D., Wise, M., Boccaccini, AR.: Thermal plasma technology for the treatment of wastes: a critical review. *J. of Hazard. Mater.* **161**, 614-626 (2009)
69. Nema, SK., Ganeshprasad, KS.: (2002) Plasma pyrolysis of medical waste. *Curr. Sci* **83**(3), 271-278 (2002)

70. Hetland, J., Lynam, S., Santen, S.: Advanced waste-to-energy technology offering safe and inert rendering of residues with almost complete volume reduction. 100%RES – A challenge for island sustainable development. 283-293 ISBN 84-609-6326-8 (2005)
72. Murphy, AB., Farmer, AJD., Horrigan, EC., McAllister, T.: Plasma Destruction of Ozone Depleting Substances, Plasma Chem. and Plasma Process **22**(3), 371-385, 10.1023/A:1015365032020 (2002)
73. Huang, H., Tang, L.: Treatment of organic waste using thermal plasma pyrolysis technology. Energ. Convers. and Manag. **48**, 1331-1337 (2007)
74. Tang, L., Huang, H., Zhao, Z., Wu, CZ., Chen, Y.: Pyrolysis of polypropylene in a nitrogen plasma reactor. Ind. Eng. Chem. Res. **42**, 1145-1150 (2003)
75. Tang, L., Huang, H., Hao, H., Wang, H., Wang, Y., Zhu, C., Li, Q.: Waste Tires Disposal by Thermal Plasma for Power Generation and Carbon Black Recovery. Power and Energ. Eng. Conf. APPEEC 2009. Asia-Pacific 27-31 March 2009:1-4 (2009)
76. Zhao, Z., Huang, H., Wu, C., Li, H., Chen, Y.: Biomass pyrolysis in an argon/hydrogen plasma reactor. Eng. Life Sci. **1** **5**, 197-199 (2001)
77. Shie, JL., Chang, CY., Tu, WK., Yang, YC., Liao, JK., Tzeng, CC., Li, HY., Yu, YJ., Kuo, CH., Chang, LC.: Major products obtained from plasma torch pyrolysis of sunflower-oil cake. Energ. & Fuels **22**, 75-82 (2008)
78. Guénadou, D., Lorcet, H., Poulain, S., Peybernès, J.: Optimisation of the working parameters for gasification of a bio-oil in a thermal plasma. Proceedings ISPC19 - 2009, Bochum, Germany – P3.16.03 (2009)
71. Van Oost, G., Hrabovsky, M., Kopecky, V., Konrad, M., Hlina, M., Kavka, T., Chumak, A., Beeckman, E., Verstraeten, J.: Pyrolysis of waste using a hybrid argon-water stabilized torch. Vacuum **80**, 1132-1137 (2006)
79. Van Oost, G., Hrabovsky, M., Kopecky, V., Konrad, M., Hlina, M., Kavka, T.: Pyrolysis/gasification of biomass for synthetic fuel production using a hybrid gas-water stabilized plasma torch. Vacuum **83**, 209-212 (2009)
80. Hlína, M., Hrabovský, M., Kopecký, V., Konrád, M., Kavka, T.: Plasma gasification of wood and production of gas with low content of tar. Czechoslovak J. of Phys. **56**(B), 1179-1184 (2006)
81. Hrabovsky, M., Konrad, M., Kopecky, V., Hlina, M., Kavka, T.: Gasification of biomass in water/gas-stabilized plasma for syngas production. Czechoslovak Journal of Physics **56**(B), 1199-1206 (2006)
82. Hrabovsky, M., Kopecký, V., Sember, V., Kavka, T., Chumak, O., Konrad, M.: Properties of Hybrid Water/gas DC arc plasma torch. IEEE Transactions on Plasma Sci. **34**(4), 1566-1575 (2006)
83. Sforza, T., Charlier, C., Hrabovsky, M., Chumak, O.: Thermal plasma assisted gasification of polymers. WDS'09 Proc. of Contributed pap., Part II:176-181 – ISBN 978-80-7378-102-6 (2009)
84. Hrabovsky, M.: Thermal plasma generators with water stabilized arc. The Open Plasma Phys. J. **2**, 99-104 (2009)
85. Tendler, M., Rutberg, P., Van Oost, G.: Plasma based waste treatment and energy production. Plasma Phys. Control. Fusion **47**, A219-A230 (2005)
86. Bratsev, AN., Popov, VE., Rutberg, AF., Shtengel', SV.: A facility for plasma gasification of waste of various types. High temperature **44**(6), 823-828 (2006)

87. Kalinenko, RA., Kuznetsov, AP., Levitsky, AA., Messerle, VE., Mirokhin, YA., Polak, LS., Sakipov, ZB., Ustimenko, AB.: Pulverized coal plasma gasification. *Plasma Chem. and Plasma Process.* **13**(1), 141-167 (1993)
88. Heberlein, J., Murphy, AB.: Thermal plasma waste treatment. *J. of Phys. D: Appl. Phys.* **41**, 053001 doi:10.1088/0022-3727/41/5/053001 (2008)
89. Bui, T., Loof, R., Bhattacharya, SC.: Multi-stage reactor for thermal gasification of wood. *Energy* **19**(4), 397-404 (1994)
90. Popov, VE., Bratsev, AN., Kuznetsov, VA., Shtengel, SV., Ufimtsev, AA.: Plasma gasification of waste as a method of energy saving. 11<sup>th</sup> European Conference on High-Technology Plasma Processes – HTTP 11 – *Journal of Physics: Conference Series* **275**, 012015 doi:10.1088/1742-6596/275/1/012015 (2011)
91. Leal-Quirós, E.: Plasma processing of municipal solid waste. *Brazilian J. of Phys.* **34**(4B), 1587-1593 (2004)
92. Willis, KP., Osada, S., Willerton, KL.: Plasma gasification: lessons learned at Ecovalley WTE facility. Proceedings of the 18th Annual North American Waste-to-Energy Conference, Orland, Florida, USA May 11-13 – NAWTEC 18 – 3515 (2010)
93. Zhang, Q., Dor, L., Fenigshtein, D., Yang, W., Blasiak, W.: Gasification of municipal solid waste in the plasma gasification melting process. *Appl. Energ.* **90**, 106-112 (2012)
94. Carabin, P., Holcroft, G.: Plasma resource recovery technology – converting waste to energy and valuable products. 13th North American Waste to Energy Conference – Orlando, Florida, USA, May 23-25 – NAWTEC 13 – 3155 (2005)
95. Byun, Y., Cho, M., Chung, JW., Namkung, W., Lee, HD., Jang, SD., Kim, YS., Lee, JH., Lee, CR., Hwang, SM.: Hydrogen recovery from the thermal plasma gasification of solid waste. *J. of Hazard. Mater.* **190**, 317-323 (2011)
96. Dave, PN., Joshi, AK.: Plasma pyrolysis and gasification of plastics waste – a review. *J. of Scientific & Industrial Res.* **69**, 177-179 (2010)
97. Park, HS., Kim, CG., Kim, SJ.: Characteristics of PE gasification by steam plasma. *J. Ind. Eng. Chem.* **12**(2), 216-223 (2006)
98. Kim, SW., Park, HS., Kim, HJ.: 100 kW steam plasma process for treatment of PCBs (polychlorinated biphenyls) waste. *Vacuum* **70**, 59-66 (2003)
99. Moustakas, K., Fatta, D., Malamis, S., Haralambous, K., Loizidou, M.: Demonstration plasma gasification/vitrification system for effective hazardous waste treatment. *J. of Hazard. Mater. B* **123**, 120-126 (2005)
100. Moustakas, K., Xydis, G., Malamis, S., Haralambous, KJ., Loizidou, M.: Analysis of results from the operation of a pilot plasma gasification/vitrification unit for optimizing its performance. *J. of Hazard. Mater.* **151**, 473-480 (2008)
101. Boudesocque, N., Lemort, F., Girold, C., Lafon, C., Vandensteendam C., Baronnet JM.: Organic liquid waste decontamination and gasification by underwater thermal plasma. 18<sup>th</sup> International Symposium of Plasma Chemistry, August 26-31, 2007b, Kyoto, Japan (2007)
102. Lupa, CJ., Wylie, SR., Shaw, A., Al-Shamma'a, A., Sweetman, AJ., Herbert, BMJ.: Experimental analysis of biomass pyrolysis using microwave-induced plasma. *Fuel Process. Technol.* **97**, 79-84 (2012)

103. Shie, JL., Tsou, FJ., Lin, KL.: Steam plasmatron gasification of distillers grains residue from ethanol production. *Bioresource Technol.* **101**, 5571-5577 (2010)
104. Shie, JL., Tsou, FJ., Lin, KL., Chang, CY.: Bioenergy and products from thermal pyrolysis of rice straw using plasma torch. *Bioresource Technol.* **101**, 761-768 (2010)
105. Rutberg, PhG., Lukyanov, SA., Kiselev, AA., Kushev, SA., Nakonechny, GhV., Nikonov, AV., Popov, SD., Serba, EO., Spodobin, VA., Surov, AV.: Investigation of parameters of the three phase high-voltage alternating current plasma generator with power up to 100 kW working on steam. 11th European Conference on High-Technology Plasma Processes (HTPP 11) – J. of Phys.: Conference Series **275**, 012006 – DOI: 10.1088/1742-6596/275/1/012006 (2011)
106. Rutberg, PhG., Safronov, AA., Popov, SD., Surov, AV., Nakonechny, GhV.: Multiphase stationary plasma generators working on oxidizing media. *Plasma Phys. Control. Fusion* **47**, 1681-1696 (2005)
107. Rutberg, PhG., Safronov, AA., Popov, SD., Surov, AV., Nakonechny, GV.: Multiphase electric-arc AC plasma generators for plasma technologies. *High Temperature* **44**(2), 199-205 (2006)
108. Rutberg, PhG., Kumkova, II., Kuznetsov, VE., Popov, SD., Rutberg, APh., Safronov, AA., Shiryayev, VN., Surov, AV.: High-voltage plasma generators of alternating current with rod electrodes stationary operating on oxidizing media. 16th IEEE Intern. Pulsed Power Conf., June 17-22. **2**, 1556-1559 (2007)
109. Kovshechnikov, VB., Antonov, GG., Ufimtsev, AA.: On the efficiency of gas heating by a three-phase AC gliding arc plasma generator. *J. of Eng. Phys. and Thermophys.* **80**(6), 1124-1129 (2007)
110. Rutberg, PhG., Aleksei, A., Safronov, A., Goryachev, VL.: Strong-current arc discharges of alternating current. *IEEE Transactions on Plasma Sci.* **26**(4), 1297-1306 (1998)
111. Rutberg, PhG., Safronov, AA., Surov, AV., Popov, SD.: Alternating current electric arc plasma generators. 28th ICPIG, Prague, Czech Republic, July 15-20 Topic **11**, 1774-1777 (2007)
112. Rutberg, PhG., Kuznetsov, VA., Bratsev, AN., Popov, VE., Shtengel', SV., Ufimtsev, AA.: Use of carbon dioxide in the chemical synthesis technologies, plasma gasification and carbon production. Symposium A, E-MRS 2010 Fall Meeting, IOP Conf. Series: Mater. Sci. and Eng. **19**, 012003 DOI:10.1088/1757-899X/19/1/012003 (2010)
113. Brothier, M., Gramondi, P., Poletiko, C., Michon, U., Labrot, M., Hacala, A.: Biofuel and hydrogen production from biomass gasification by use of thermal plasma. *High Temp. Mater. Process* **11**, 231-244 (2007)
114. Westinghouse Plasma Corporation official web site. <http://www.westinghouse-plasma.com>. Accessed 05 June 2012
115. Europlasma official web site. <http://www.europlasma.com>. Accessed 05 June 2012
116. Phoenix Solutions Company official web site. <http://www.phoenixsolutionsco.com>. Accessed 12 Sept 2012
117. Morrin, S., Lettieri, P., Chapman, C., Mazzei, L.: Two stage fluid bed-plasma gasification process for solid waste valorisation: technical review and preliminary thermodynamic modelling of sulphur emissions. *Waste Manag.* **32**, 676-684 (2012)
118. Tetronics International official web site. <http://www.tetronics.com>. Accessed 25 June 2012
119. Advances Plasma Power official web site. <http://www.advancedplasmamapower.com>. Accessed 25 June 2012
120. Alter NRG Corp. official web site. <http://www.alternrg.com>. Accessed 25 June 2012
121. C.H.O-Power official web site. <http://www.cho-power.com>. Accessed 06 June 2012

122. Plasma Arc Technologies Inc. official web site. <http://www.plasmaarctech.com>. Accessed 04 June 2012
123. Plasco Energy Group Inc. official web site. <http://www.plascoenergygroup.com>. Accessed 05 June 2012
124. EnerSol Technologies Inc. official web site. <http://www.enersoltech.com>. Accessed 25 June 2012
125. Bellwether Gasification Technologies Ltd. official web site. <http://www.bgt-online.com>. Accessed 12 Sept 2012
126. STARTECH'S Plasma Converter System official web site. <http://theplasmasolution.com>. Accessed 12 Sept 2012
127. Green Power Systems official web site. <http://www.greenpowersystems.com>. Accessed 25 June 2012
128. PEAT International official web site. <http://www.peat.com>. Accessed 25 June 2012
129. InEnTec Inc. official web site. <http://www.inentec.com/>. Accessed 25 June 2012
130. Pyrogenesis Canada Inc. official web site. <http://www.pyrogenesis.com>. Accessed 04 June 2012
131. Inertam official web site. <http://www.inertam.com>. Accessed 06 June 2012
132. Messerle VE, Peregudov VS (1995) Ignition and stabilization of combustion of pulverized coal fuels by thermal plasma. Vol. 2 : Investigation and design of thermal plasma technology / Cambridge Interscience Publishing, London, 323-343
133. Serbin, SI., Matveev, IB.: Theoretical investigations of the working processes in a plasma coal gasification system. *IEEE Transactions on Plasma Sci.* **38**(12), 3300-3305 (2010)
134. Matveev, IB., Serbin, SI.: Modeling of the coal gasification processes in a hybrid plasma torch. *IEEE Transactions on Plasma Sci.* **35**(6), 1639-1647 (2007)
135. Matveev, IB., Serbin, SI.: Theoretical and experimental investigations of the plasma-assisted combustion and reformation system. *IEEE Transactions on Plasma Sci.* **38**(12), 3306-3312 (2010)
136. Matveev, IB., Serbin, SI., Lux, SM.: Efficiency of a hybrid-type plasma assisted fuel reformation system. *IEEE Transactions on Plasma Sci.* **36**(6), 2940-2946 (2008)
137. Matveev, IB.: Plasma or retirement. Alternatives to the coal-fired power plants. *IEEE Transactions on Plasma Sci.* **39**(12), 3259-3262 (2011)
138. Matveev, I., Serbin, S.: Preliminary Design and CFD Modeling of a 1 MW Hybrid Plasma Torch for Waste Destruction and Coal Gasification, 2nd Int. Workshop and Exhibition on Plasma Assisted Combustion, Falls Church, Virginia 43–44 (2006)
139. Zverev, SG., Frolov, VYa., Ivanov, DV., Ushin, BA., Petrov, GK., Matveev, IB.: Experimental Investigations of the Hybrid Plasma Torch with Reverse Vortex Stabilization , 4th Int. Workshop and Exhibition on Plasma Assisted Combustion (IWEPAC), Falls Church, Virginia, 2008 39–44 (2008)
140. Matveev, I., Matveyeva, S., Kirchuk, S., Zverev, S.: Development and experimental investigations of high power hybrid plasma torches – 5th International Workshop and Exhibition on Plasma Assisted Combustion (IWEPAC), Alexandria, Virginia, USA 66–68 (2009)
141. Applied Plasma Technologies official web site. <http://www.plasmacombustion.com>. Accessed 16 March 2012
142. Fabry, F., Flamant, G., Fulcheri, L.: Carbon black processing by thermal plasma. Analysis of the particle formation mechanism. *Chem. Eng. Sci.* **56**, 2123-2132 (2001)
143. Fulcheri, L., Probst, N., Flamant, G., Fabry, F., Grivei, E., Bourrat, X.: Plasma processing: a step towards the production of new grades of carbon black. *Carbon* **40**, 169-176 (2002)

144. Ravary, B., Fulcheri, L., Bakken, JA., Flamant, G., Fabry, F.: Influence of the Electromagnetic forces on momentum and heat transfer in a 3-Phase AC plasma reactor. *Plasma Chem. and Plasma Process.* **19**(1), 69-89 (1999)
145. Ravary, B., Fulcheri, L., Flamant, G., Fabry, F.: Analysis of a 3-phase A.C. plasma system. *High Temp. Mater. Process.* **2**, 245-260 (1998)
146. Enviroparks Ltd. official web site. <http://www.enviroparks.co.uk/>. Accessed 25 June 2012
147. Sunbay Energy, Ontario Gasification Facility, Port Hope, Ontario. <https://porthope.civicweb.net/content/pdfstorage/F89DEE77D95A41D3BD4BE21E38103EF4-Delegation%20Apr%2015%2008%20Sunbay%20OGF.pdf>. Accessed 17 Sept 2012
148. Carabin, P., Gagnon, JR.: Plasma gasification and vitrification of ash-conversion of ash into glass-like products and syngas. 2007 World of Coal Ash (WOCA) May 7-10 2007 Covington, Kentucky, USA (2007)
149. Kaldas, A., Carabin, P., Picard, I., Chevallier, P., Holcroft, G.: Treatment of ship sludge oil using a plasma arc waste destruction system (PAWDS). IT3'07 Conference, May 14-18, Phoenix, AZ (2007)
150. Kaldas, A., Picard, I., Chronopoulos, C., Chevalier, P., Carabin, P., Holcroft, G., Alexander, G., Spezio, J., Mann, J., Molintas, H.: Plasma arc waste destruction system (PAWDS) A novel approach to waste elimination aboard ships. MEETS06 – Crystal City (Arlington), Virginia, January 23 – 25 (2006)
151. Ojha, A., Reuben, AC., Sharma, D.: Solid waste management in developing countries through plasma arc gasification- an alternative approach. ICESD 2012: 5-7 January 2012, Hong Kong, APCBEE Procedia **1**, 193-198 (2012)
152. Clark, BJ., Rogoff, MJ.: Economic feasibility of a plasma arc gasification plant, city of Marion, Iowa. Proceedings of the 18th Annual North American Waste-to-Energy Conference, Orland, Florida, USA May 11-13 – NAWTEC 18 – 3502 (2010)
153. Ducharme, C., Themelis, N.: Analysis of thermal plasma – assisted waste-to energy processes. Proceedings of the 18th Annual North American Waste-to-Energy Conference, Orland, Florida, USA May 11-13 – NAWTEC 18 – 3582 (2010)
154. Yang, L., Wang, H., Wang, H., Wang, D., Wang, Y.: Solid waste plasma disposal plant. *J. of Electrostatics* **69**, 411-413 (2011)
155. Pourali, M.: Application of plasma gasification technology in waste to energy – challenges and opportunities. *IEEE Transactions on Sustain. Energ.* **1**(3), 125-130 (2010)
156. Artemov, AV., Bul'ba, VA., Voshchinin, SA., Krutyakov, YA., Kudrinskii, AA., Ostryi, II., Pereslavl'tsev, AV.: Technical and economic operation parameters of a high-temperature plasma plant for production and consumption waste conversion (calculation results and their analysis). *Russian J. of Gen. Chem.* **82**(4), 808-814 (2012)
157. Loghin, I.: Market barriers to the integrated plasma gasification combined cycle plant implementation – Romanian case. *U.P.B. Sci. Bull., Series C* **70**(2), 111-120 (2008)
158. Seiler, JM., Hohwiller, C., Imbach, J., Luciani, JF.: Technical and economical evaluation of enhanced biomass to liquid fuel processes. *Energy* **35**, 3587-3592 (2010)
159. Byun, Y., Namkung, W., Cho, M., Chung, JW., Kim, YS., Lee, JH., Lee, CR., Hwang, SM.: Demonstration of thermal plasma gasification/vitrification for municipal solid waste treatment. *Environ. Sci. Technol.* **44**, 6680-6684 (2010)



160. Shie, J.L., Chang, C.Y., Chen, C.S., Shaw, D.G., Chen, Y.H., Kuan, W.H., Ma, H.K.: Energy life cycle assessment of rice straw bio-energy derived from potential gasification technologies. *Bioresource Technol.* **102**, 6735-6741 (2011)

## List of Tables

**Table 1** Main chemical reactions of gasification

N°	Reaction name	Chemical reaction	Reaction enthalpy $\Delta H^{(1)}$
(1)	$C_nH_mO_k$ partial oxidation	$C_nH_m + \frac{n}{2} O_2 \leftrightarrow \frac{m}{2} H_2 + n CO$	Exothermic
(2)	Steam reforming	$C_nH_m + n H_2O \leftrightarrow (n + \frac{m}{2}) H_2 + n CO$	Endothermic
(3)	Dry reforming	$C_nH_m + n CO_2 \leftrightarrow \frac{m}{2} H_2 + 2n CO$	Endothermic
(4)	Carbon oxidation	$C + O_2 \rightarrow CO_2$	-393.65 kJ.mol <sup>-1</sup>
(5)	Carbon Partial oxidation	$C + \frac{1}{2} O_2 \rightarrow CO$	-110.56 kJ.mol <sup>-1</sup>
(6)	Water-gas reaction	$C + H_2O \leftrightarrow CO + H_2$	+131.2 kJ.mol <sup>-1</sup>
(7)	Boudouard reaction	$C + CO_2 \leftrightarrow 2 CO$	+172.52 kJ.mol <sup>-1</sup>
(8)	Hydrogasification	$C + 2 H_2 \leftrightarrow CH_4$	-74.87 kJ.mol <sup>-1</sup>
(9)	Carbon monoxide oxidation	$CO + \frac{1}{2} O_2 \rightarrow CO_2$	-283.01 kJ.mol <sup>-1</sup>
(10)	Hydrogen oxidation	$H_2 + \frac{1}{2} O_2 \rightarrow H_2O$	-241.09 kJ.mol <sup>-1</sup>
(11)	Water-gas shift reaction	$CO + H_2O \leftrightarrow CO_2 + H_2$	-41.18 kJ.mol <sup>-1</sup>
(12)	Methanation	$CO + 3 H_2 \leftrightarrow CH_4 + H_2O$	-206.23 kJ.mol <sup>-1</sup>

<sup>(1)</sup> T = 298 K, P = 1.013 10<sup>5</sup> Pa, carbon as solid and water in vapor form

**Table 2** Tar and solid particles rates in the gasification raw-gas in function of the reactor configuration

Reactor	Tar rates, g.Nm <sup>-3</sup>			Solid particles rates, g.Nm <sup>-3</sup>		
	Min	Max	R.R.	Min	Max	R.R.
Updraft	1	150	20 – 100	0.1	3	0.1 – 1
Downdraft	0.04	6	0.1 – 1.2	0.01	10	0.1 – 0.2
Fluidized bed	< 0.1	23	1 – 15	1	100	2 – 20

R.R.: Representative Range in which are most of the processes studied

**Table 3** Performances of plasma gasification from scientific literature and comparison versus autothermal gasification

Plasma Torch	Ref.	Waste	LHV MJ.kg <sup>-1</sup>	Oxidizing agent	Power kW	Syngas Nm <sup>3</sup> .kg <sup>-1</sup>	H <sub>2</sub> Vol%	CO Vol%	LHV MJ.Nm <sup>-3</sup>	η %	Net electrical efficiency, %
DC	[81]	Wooden Sawdust	-	CO <sub>2</sub> + O <sub>2</sub>	95 – 153	1 – 1.9	22 – 46	44 – 68	-	-	-
	[25]	Tires	37.33	H <sub>2</sub> O	35.2	-	24.12	14.17	8.96	-	-
	[74]	Polypropylene	-	H <sub>2</sub> O	35.2	2.17	27.06	13.33	-	-	-
	[94]	MSW	-	H <sub>2</sub> O + air	-	-	8 – 14	20 – 29	4.1 – 5.2	-	-
	[94]	Hazardous Waste	-	H <sub>2</sub> O + air	-	-	28.3	18.8	6.0	-	-
	[92]	MSW/tires mix	13.95	O <sub>2</sub>	3,22	1.10	22.62	40.46	11.9	79	28 – 46 <sup>#</sup>
AC	[32]	Wood residues	13.9	Air	2.16 <sup>+</sup>	2.45	28	23.6	13.5	84	29 – 49 <sup>#</sup>
	[85]	Wood	16	None	3.6 <sup>+</sup>	2.48	24.5	31.4	6.16	78	28 – 46 <sup>#</sup>
	[85]	RDF *	15	None	3.82 <sup>+</sup>	2.46	26.3	27.5	5.88	77	27 – 45 <sup>#</sup>
	[85]	Tires	33	H <sub>2</sub> O	6.66 <sup>+</sup>	5.03	30.6	24	5.89	75	26 – 44 <sup>#</sup>
Autothermal	[9]	MSW	7 – 18	Air – O <sub>2</sub>	-	-	-	-	4 – 7	50 – 80	15 – 24

\* The refused derived fuel (RDF) terms the specially prepared dry fuel consisting of chips of wood, paper, plastic, fabric, rubber and other hydrocarbons.

+ Energy in MJ for 1 kg of waste

# With electrical conversion efficiency of around 0.35 for steam thermodynamic cycle [32] and around 0.59 for the combined cycle [37]

**Table 4** Properties of the Westinghouse DC Torches

Model	Power kW	Diameter In	mm	Length In	mm	Weight lb	kg
Marc 3a	80 – 300	3.5	89	32.5 mini	826 mini	27	12.2
Marc 3HC	5 – 150	3.5	89	20.2 mini	513 mini	16.6	7.5
Marc 11L	300 – 800	18	457	35	889	450	204
Marc 11H	700 – 2400	18	457	35	889	450	204

**Table 5** Main technical characteristics of the Europlasma DC Torches

Model	Power, kW	Plasma Gas
Hot Cathode	25 – 100	Ar, He, H <sub>2</sub>
Cold Cathode	100 – 300	Air, CO, CO <sub>2</sub>
Cold Cathode	300 – 800	Air
Cold Cathode	800 – 2000	Air
Cold Cathode	1500 – 4000	Air

**Table 6** Main technical characteristics of the PSC DC Torches

Model	Power, kW
PT50	10 – 100
PT150	100 – 300
PT200	200 – 900
PT250	800 – 3000
PT255	1500 – 3000

**Table 7** Technical specifications of the 3-phase AC power supply

Input	380V 50hz 3-phase
Max. Output Voltage	0-500 V 3-phase
Max. Output Current	0-400 A
Max. Output Power	263 kVA
Output frequency	84, 168, 338 or 675 Hz

**Table 8** Main plants for waste gasification by plasma currently in operation around the world and plant projects for the next years [114-116, 119-131, 147-148]

Location	Raw material	Capacity (TPD)	Start Date	Production	Plasma Technology	Company
Mihama-Mikata, Japan	MSW/WW Sludge	25	2002		DC, Westinghouse	Hitachi Metals Ltd.
Utashinai, Japan	MSW/ASR	300	2002		DC, Westinghouse	Hitachi Metals Ltd.
Yoshi, Japan	MSW	151	1999		DC, Westinghouse	Hitachi Metals Ltd.
Pune, India	Hazardous Waste	68	2009	1,6 MWe	DC, Westinghouse	Maharashtra Env. Pow.
Nagpur, India	Hazardous Waste	68	2010	1,6 MWe	DC, Westinghouse	Maharashtra Env. Pow.
Shanghai, China			Project		DC, Westinghouse	
Tallahassee, USA	MSW	910	Project	35 MWe	DC, Westinghouse	Green Power Systems
Morcenx, France	Industrial/Biomass	137	2012	12 MWe	DC, Europlasma	CHO-Power
Hull, Sunderlan, Barry & Barrow, UK	Industrial/Biomass	107 x 4 plants	Project	37.5 MWe	DC, Europlasma	CHO-Power
Port Hope, Canada	MSW/TDF	400	Project	26 MWe	DC, Europlasma	Sunbay Energy Corp.
Hirwaun, UK	MSW/industrial	~750	Project (2015)	20 MWe	DC, Europlasma	EnviroParks Limited
Ottawa, Canada	MSW	85	Demonstration facility	1 MWe/ton	DC, PSC	Plasco Energy Group Inc.
Trail Road, USA	MSW		Demonstration facility	0.88 MWe/ton	DC, PSC	Plasco Energy Group Inc.
Los Angeles, USA	MSW		Project		DC, PSC	Plasco Energy Group Inc.
Beijing, China	MSW	200	Project		DC, PSC	Plasco Energy Group Inc.
Tainan City, Taiwan	Hazardous Waste	3-5	2005		DC, Homemade	PEAT International
Iizuka, Japan	Industrial Wastes	10	2004		DC, Homemade	InEnTec
U.S. Navy	Shipboard Wastes	7	2004		DC, Pyrogenesis	Pyrogenesis
Hurlburt Field, USA	MSW/Hazardous	10.5	2011		DC, Pyrogenesis	Pyrogenesis
Faringdon, UK			Demonstration facility		DC, Tetronics	Advanced Plasma Power
Swindon	MSW	91,000 t/year	2008	16.3 MWe	DC, Tetronics	Advanced Plasma Power
South Wales	MSW		Project		DC, Tetronics	Advanced Plasma Power
North of England	MSW		Project		DC, Tetronics	Advanced Plasma Power
South West England	MSW		Project	17 MWe	DC, Tetronics	Advanced Plasma Power
Scotland – East Coast	MSW	91,000 t/year	Project		DC, Tetronics	Advanced Plasma Power
Brazil	MSW		Project		DC, Tetronics	Advanced Plasma Power
Brazil	ASR		Project		DC, Tetronics	Advanced Plasma Power
Belgium	Landfill	246 x 5 plants	Project	100 MWe	DC, Tetronics	Advanced Plasma Power
Swindon	Residual wastes		Demonstration facility	BSNG	DC, Tetronics	Advanced Plasma Power

ASR: Auto Shredder Residue, WW Sludge: Waste Water Sludge, TDF: Tire Derived Fuel, BSNG: Bio Substitute Natural Gas, TPD: Metric Tons Per Day

**Table 9** Waste gasification project using Alter NRG's plasma technology at various stages of development (Q1 2011) [120]

COUNTRY	PRODUCTION	Site Selection	Feasibility Study	Feedstock Agreement	Initial Engineering	Regulatory Application	Regulatory Approval	Financing	Detailed Engineering	Construction,	Commissioning / Start-up	Operation
<b>NORTH AMERICA</b>												
SE, US	Biomass-to-Ethanol											
St. Lucie, FL	WTE											
Atlantic City, NJ	WTE											
Milwaukee, WI	WTE											
Ontario, Canada	WTE											
Minnesota	WTE repowering											
Madison, PA	Biomass-to-Ethanol											
US – Strategic Licensor	WTE (3 projects)											
<b>EUROPEAN UNION</b>												
Poland	WTE											
Spain	WTE											
United Kingdom	WTE											
Spain	Industrial/hazardous											
Italy	Medical Waste											
<b>INDIA</b>												
India	Hazardous WTE (3-5 proposed facilities)											
Pune	Hazardous WTE											
Nagpur	Hazardous WTE											
<b>CHINA</b>												
Central China	Biomass-to-Ethanol (150 known projects)											
Western China	WTE											
Central China	WTE											
Southern China	WTE (2-5 projects – various stages)											
<b>AUSTRALIA</b>												
Melbourne	Waste-to-ethanol											
Geelong	Waste-to-energy											
Kwinana	Waste-to-energy											
<b>RUSSIA</b>												
Moscow	WTE (5 projects)											